

### WEST BENGAL STATE UNIVERSITY

B.Sc. Honours 6th Semester Examination, 2024

## CEMADSE04T-CHEMISTRY (DSE3/4)

#### GREEN CHEMISTRY

Time Allotted: 2 Hours

Full Marks: 40

The figures in the margin indicate full marks. Candidates are required to give their answers in their own words as far as practicable. All symbols are of usual significance.

#### Answer any three questions taking one from each Group

#### **GROUP-A**

- 1. (a) Write the relation between mass intensity (MI) and E-factor. What is the value of 2+1E-factor for an ideal process? (b) Define the term 'atom economy'. 'Addition reactions are 100% atom economical 2+2reactions.' — Justify the statement. (c) Why microwave is a non-ionizing radiation? Give an example of ionizing radiation. 1+1+2What are the differences between microwave heating and conventional heating? (d) What is sonoluminescence? Why are monoatomic gases more prone towards 2+1sonoluminescence? 1 + 3(e) What is cavitation? Name different types of cavitation and specify which type is mainly responsible for chemical reaction. (f) What are susceptors in MW-induced reactions? Mention two differences between 2+2susceptors and catalysts. 2. (a) Discuss life time of a chemical product and use of safer substances following the 3 principles of green chemistry. 5 (b) Which of the following reactions has higher atom economy? Explain your answer.
  - Reaction 1:

$$CH_{3}CH=CH_{2} \xrightarrow{Cl_{2}/H_{2}O} CH_{3}CHOHCH_{2}Cl \xrightarrow{NaOH} O \xrightarrow{H_{2}O, 120-190^{\circ}C} CH_{3}CHOHCH_{2}OH$$

Reaction 2:

$$\text{HOCH}_2\text{CHOHCH}_2\text{OH} + \text{H}_2 \xrightarrow{\text{CuCrO}_4, 200^{\circ}\text{C}} \text{CH}_3\text{CHOHCH}_2\text{OH}$$

(c) What do you mean by ISD? Mention the use and green synthesis of carbaryl. 1+1+2

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	(d)	Give one example of photocatalyst.	
		What are the factors involved in measuring toxic effect of chemical substance? Explain with examples.	3
	(f)	What is ionic liquid (IL)? Give an example. What is the role of fluorous biphasic solvent in green synthesis?	2+1+3
		GROUP-B	
3.	. (a)	Why oxidation of cyclohexane or cyclohexanol to adipic acid is not considered as green synthesis? Write down one green approach for the synthesis of adipic acid.	2+3
	(b)	What are the differences between pigments and dyes?	. 3
		Why organic dyes are light sensitive?	2
4		What are enzymatic inter-esterification reactions? Describe how enzymatic inter- esterification helps for production of no trans-fats and oils.	2+2
	(b)	Briefly discuss the role of scCO <sub>2</sub> in cleaning industry. Write the name and structure of one surfactant which can form micelle in scCO <sub>2</sub> .	3+2
	(c)	) Give an example of safe marine anti-foulant.	1
		GROUP-C	
4	5. (a	Biocatalysts are different from traditional catalysts. — Explain. Give two examples of reactions where biocatalysts are used.	2+4
	(b	What is multifunctional reagent? Give an example.	2
,	6. (a	a) What are co-crystals? Briefly explain any two methods of solid state co-crystallisation.	1+4
	/L	b) What are DRAM and RRAM?	2
			1
	. (	c) Give one example of green oxidant.	